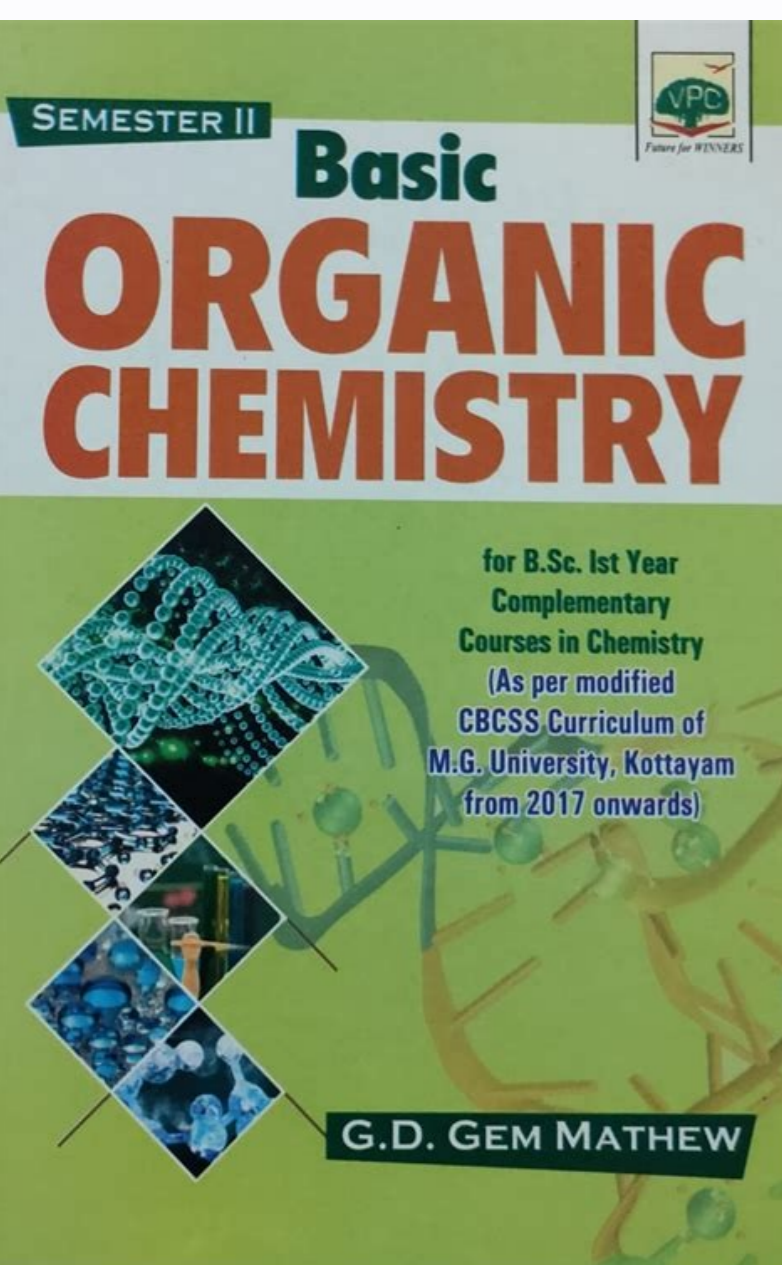


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Total No. of Questions : 5]

SEAT No. :

P89

[Total No. of Pages : 4

[4317] - 6
F.Y. B.Sc.
CHEMISTRY - II
Organic and Inorganic Chemistry
(Paper - II) (2008 Pattern) (Theory)

Time : 3 Hours]

[Max. Marks : 80

Instructions to the candidates :

- 1) All questions are compulsory.
- 2) Draw neat diagrams wherever necessary.
- 3) Figures to the right indicate full marks.

Q1) Answer the following questions : [16]

- a) Why is staggered conformation more stable than eclipsed conformation?
- b) Give applications of organic compounds in our daily life.
- c) Ethyl alcohol is water soluble but n-butyl alcohol is water insoluble. Explain.
- d) Maleic acid undergoes cyclisation at low temperature but fumaric acid does not. Explain.
- e) Benzene does not undergo addition reactions. Explain.
- f) What is oxidation number of
 - i) P in $(P_2O_5)^{4-}$
 - ii) S in $(S_4O_6)^{2-}$
- g) Alkali metals do not show +2 oxidation state. Explain.
- h) Draw the structures of following :
 - i) XeO_2F_2
 - ii) $[XeO_6]^{4-}$

Q2) a) Attempt any two of the following : [8]

- i) Discuss the acidic nature of phenol. Give any two reactions of phenol.
- ii) What is Resonance effect? Explain +R and -R effects with suitable examples.
- iii) Explain isomerism. Give classification of structural isomerism with examples.

P.T.O.

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